

# Dehydration kinetics of hydrohalite

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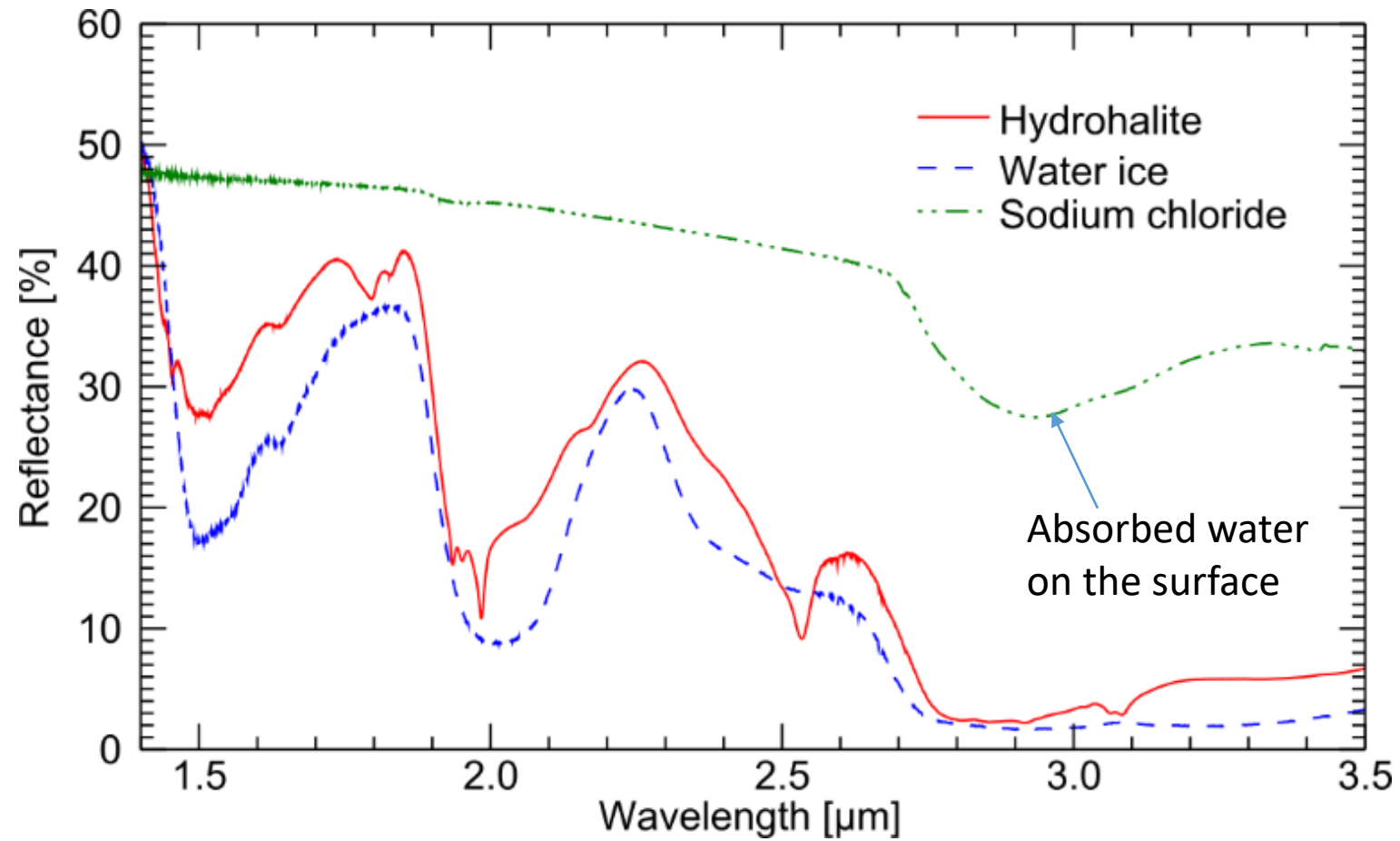
# Motivation

- Mostly driven by upcoming Europa Clipper Mission
- Infer the surface history of Europa
- Link surface composition to ocean composition and subsurface geological processes
- Link observables from NIR spectroscopy to past and present aqueous environment, i.e. subsurface ocean outflows (Miyamoto et al. 2005) or potential plume deposits (Sparks et al. 2017)
- Build up a systematic catalog of laboratory measurements, with relevant timescales

# Hydrohalite

- NaCl is one species of salt that has been proposed to exist on Europa (Kargel et al. 2000, Icarus)
- The NIR spectrum of anhydrous NaCl is flat and indistinct and therefore hard to detect from remote sensing
- Hydrohalite ( $\text{NaCl} \cdot 2\text{H}_2\text{O}$ ) forms in saturated NaCl solution below 268K
- Only stable hydrated state of sodium chloride under Europa conditions, other salts such as sodium- or magnesium-sulfates have numerous hydration states

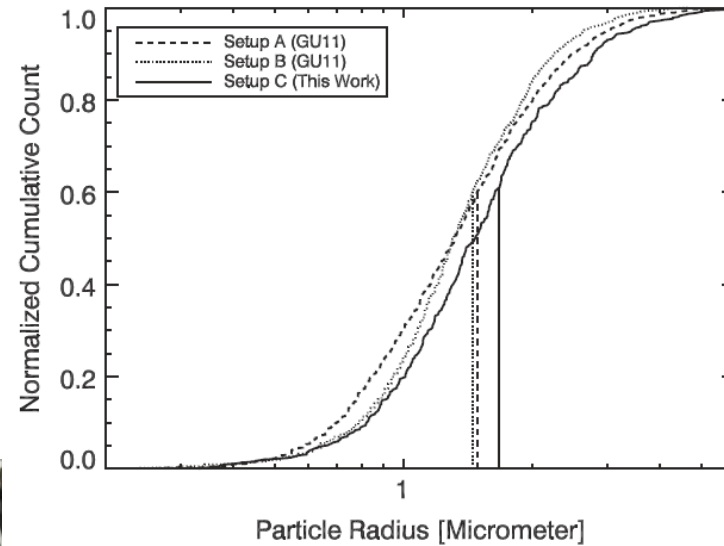
# Hydrohalite



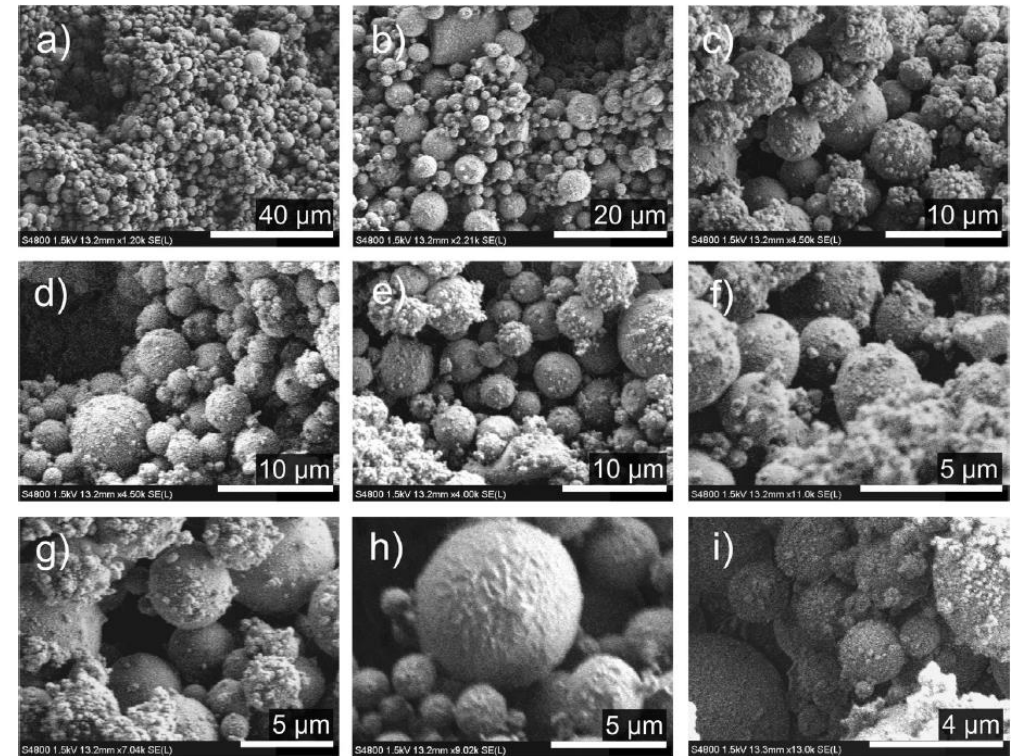
# Variable parameters

- Grains size
- UV irradiation wavelength
- Temperature
- Electron energy

# Sample production

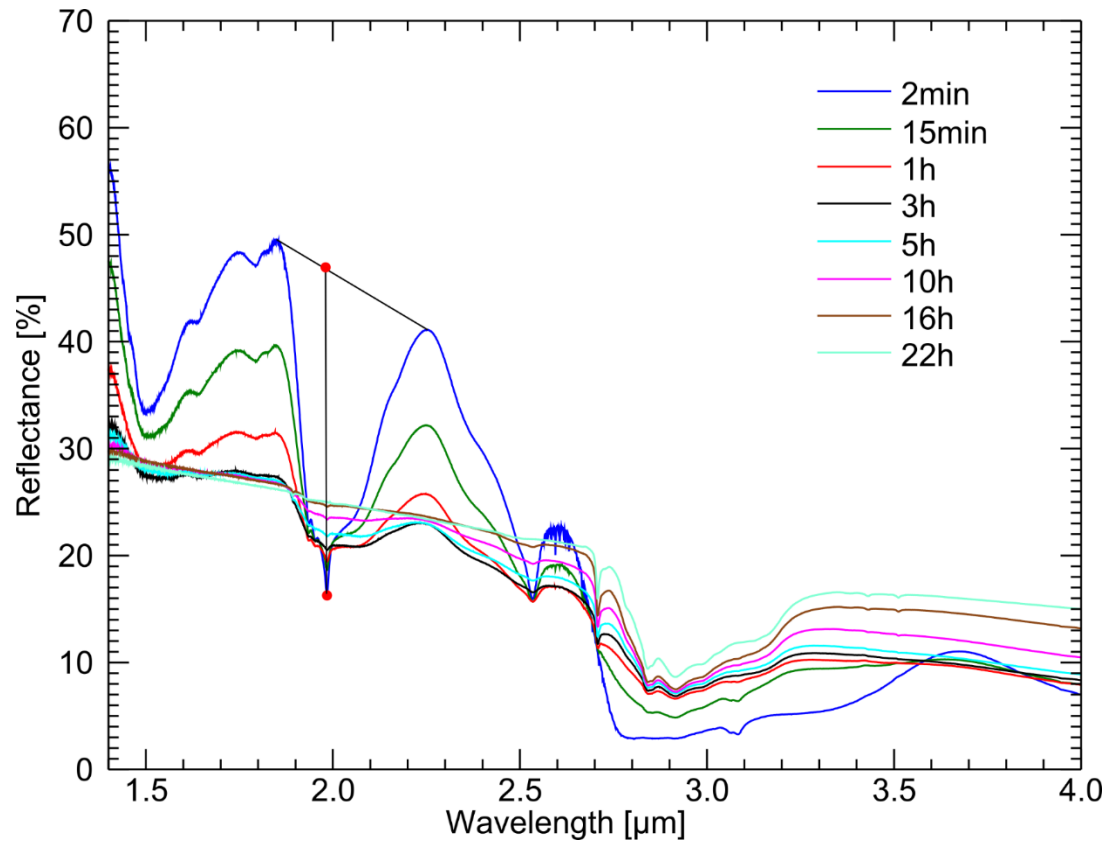


$$r = 1.47 + 0.96 - 0.58 \mu\text{m}$$

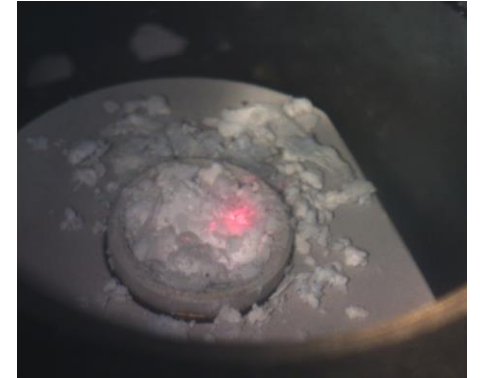
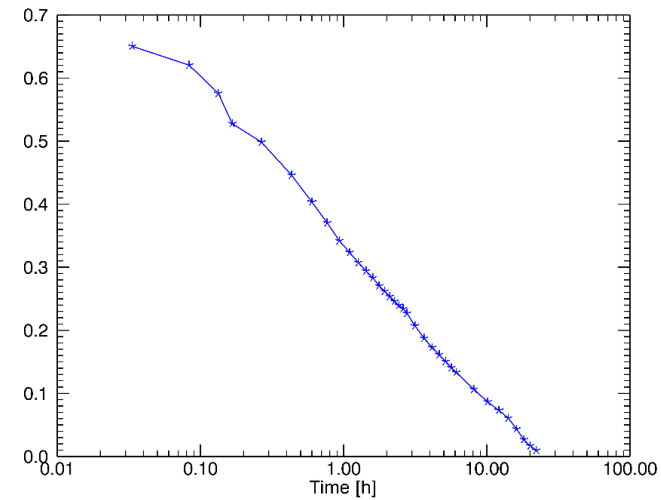
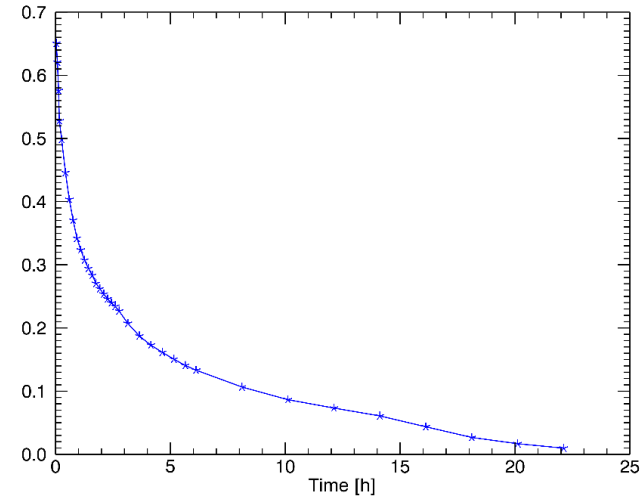


Jost et al., 2013, Icarus

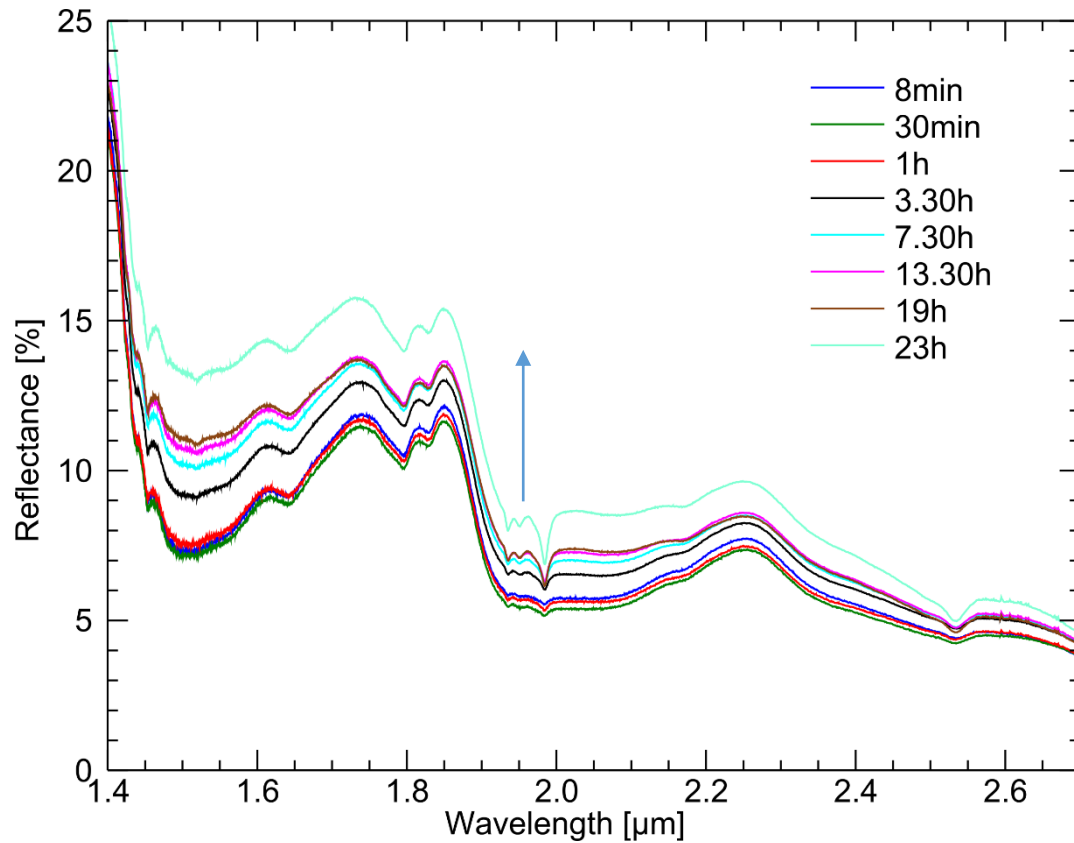
# Preliminary results



$T = -110^\circ\text{C}$ ,  $P = 2 \times 10^{-3}$  mbar

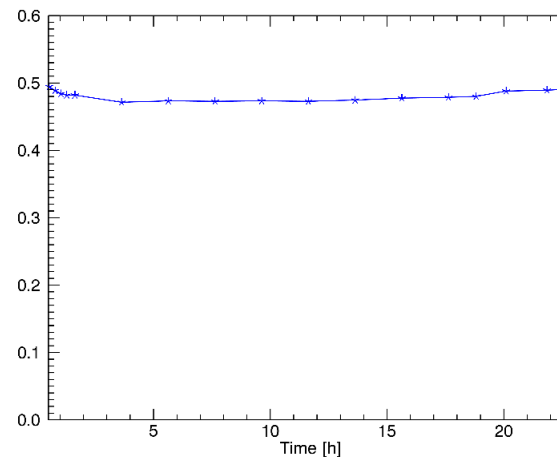
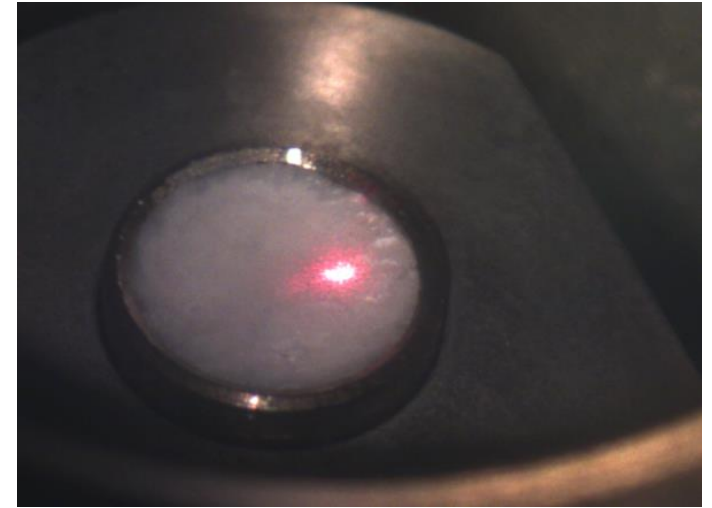


# Preliminary results



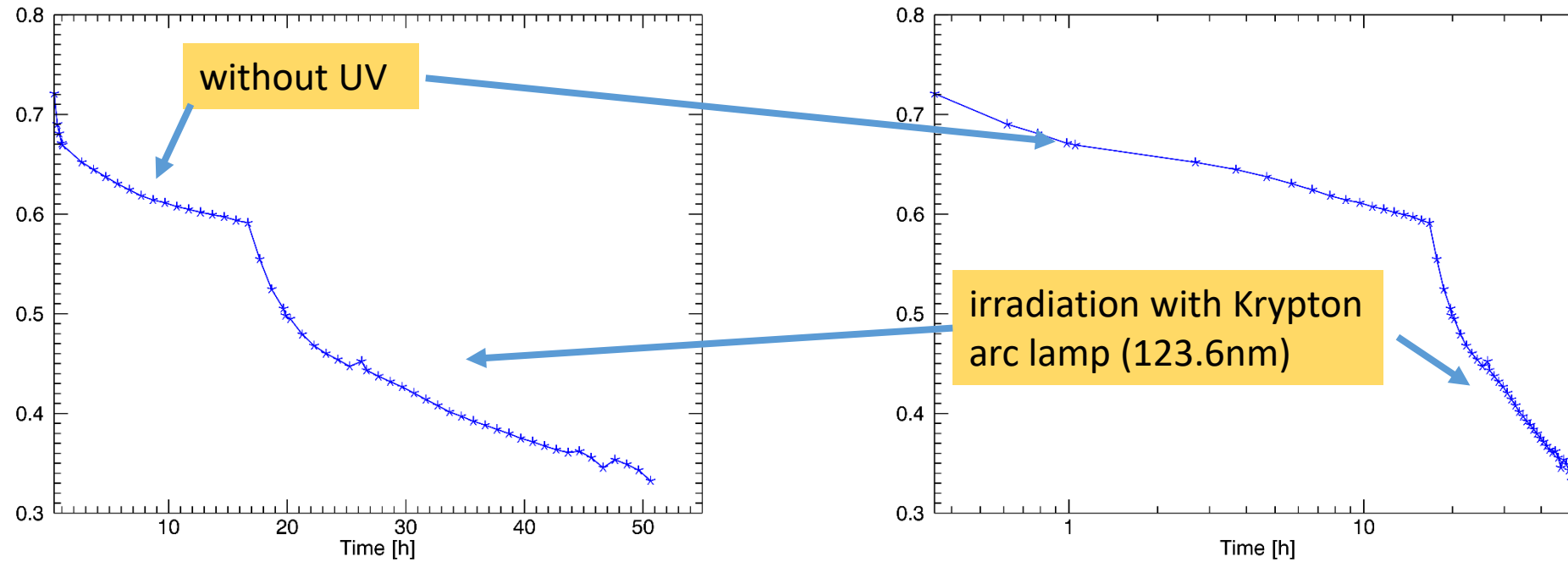
$T = -90^\circ\text{C}$ ,  $P = 2 \times 10^{-3}$  mbar

Brine frozen in  
side cryo-stage





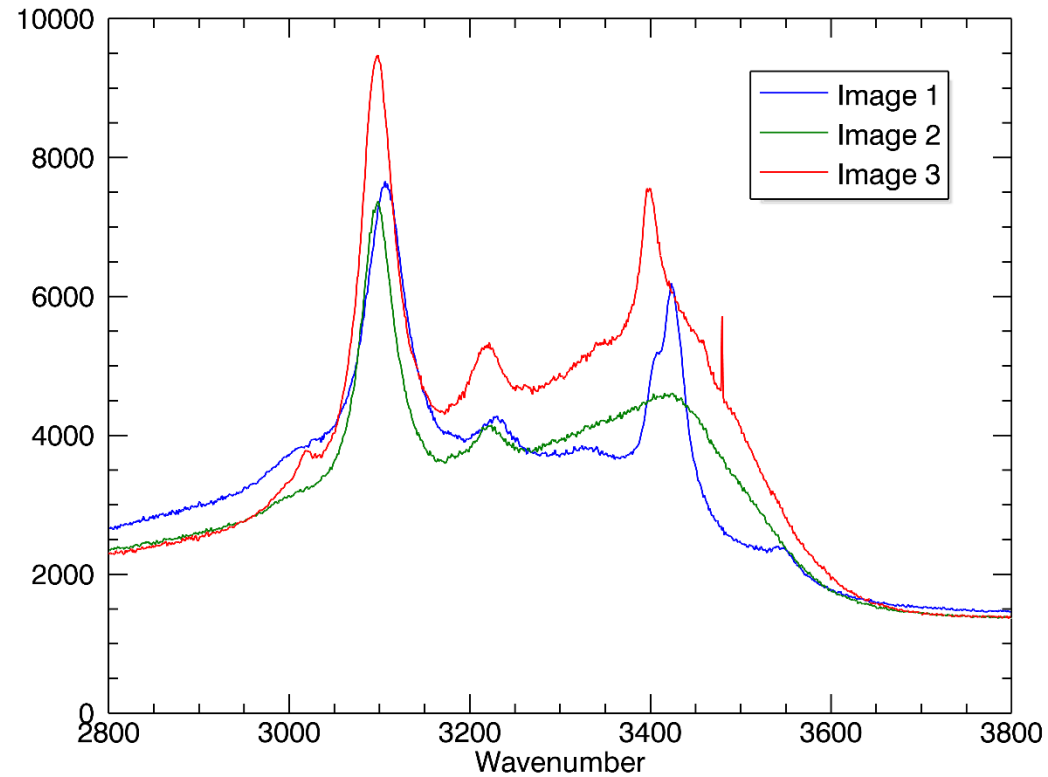
# Preliminary results



$T = -170^{\circ}\text{C}$ ,  $P = 2 \times 10^{-3}$  mbar

# Preliminary results

## Raman spectroscopy



T=-160°C, P=ambient

# Problems, work to do

- The spectral detection of Hydrohalite by Europa clipper would constitute an unambiguous
- Apply different UV arc lamp: I<sub>2</sub>, Xe, Hg

# Summary

- The spectral detection of Hydrohalite by Europa clipper would constitute an unambiguous hint for very recent aqueous activity since hydrated salt minerals on the surface dehydrate relatively quickly in high-radiation environments
- Our current laboratory investigation tries to elucidate the relevant timescales for this processes
- Preliminary results show a strong temperature dependence as well as influence from morphological structure